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study of how
fast chemical
reactions occur
and of the
factors that
affect these
rates. The study

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of reaction rates is closely related to the study of

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mechanisms,

where a reaction

mechanism is a

theory that

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the change in

the

concentration of

a reactant or a

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rate = $-\frac{d[A]}{dt}$

rate = $\frac{d[B]}{dt}$

$d[A]$ = change in

concentration of

A over time

period dt $d[B]$ =

change in

concentration of

B over time

period dt

Because $[A]$

decreases with

time, $d[A]$ is

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is guided and

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energetics, but

the actual route

it takes and the

speed with which

it occurs is the

subject of

"dynamics".

Dynamics is

itself divided

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into two general areas: kinetics, which deals with the rate of change and is the subject of this lesson.

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how to predict

their behavior,

and how to

control them. It

is closely

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very brief time
periods. It

considers state-
to-state

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product

molecules in
specific quantum

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