

Chapter 9 Review Stoichiometry Section 1 Answers

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Stoichiometry. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. _____ The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.

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Stoichiometry. SECTION 2. PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. The following equation represents a laboratory preparation for oxygen gas: 2KClO₃(s) → 2KCl(s) + 3O₂(g) ... CHAPTER 9 REVIEW ...

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SECTION 2 continued Date Class _____ 60.2 9 42.1 1 a. \ tt mash 01 ox aen Cas i pridui.ed it 100. of lithium c a C ti. I o c. i o g di I C1O c — LCi(,; — h. The oxygen gas produced in part ahas density ot 1.43 g/L iculate the olurne of thi as.. 76 STOICHIOMETRY MODERN CHEMISTRY a. —. 81 g 6. A car air bag requires 70. L of nitrogen gas ...

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